SELF-TESTS

In the matching section, there is only one answer to each question; however, the lettered options (a, b, c, etc.) may be used more than once or not at all.

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- ____ 1. Energy-yielding series of reactions.
- ____ 2. Means "whole enzyme."
- ___ 3. A nonprotein component of an active enzyme.
- 4. A measure of the rate of activity of an enzyme.
- 5. A protein portion of an enzyme, inactive without a cofactor.
- 6. A group of enzymes that function as electron carriers in respiration and photosynthesis.
- ____ 7. A mechanism by which fatty acids are degraded.

- a. Catabolism
- b. Anabolism
- c. Turnover number
- d. Apoenzyme
- e. Coenzyme
- f. Holoenzyme
- g. Beta oxidation
- h. Cytochromes

II. Matching

- ____ 1. Both the carbon source and energy source are usually the same organic compound.
- Photosynthetic, but uses organic material rather than carbon dioxide as a carbon source.
- 3. The photosynthetic purple nonsulfur bacteria would be classified in this nutritional group.
- 4. Photosynthetic bacteria that use carbon dioxide as a carbon source.
- ____ 5. Changes the shape of the active site of an enzyme.
- 6. Very similar in shape or chemistry to the normal enzyme substrate.

- a. Competitive inhibitor
- b. Noncompetitive inhibitor
- c. Photoautotroph
- d. Chemoautotroph
- e. Photoheterotroph
- f. Chemoheterotroph

III. Matching	
1. Hexose monophosphate shunt.	a. Fermentation
2. The final electron acceptor is oxygen.	b. Glycolysis
 3. Produces important intermediates that act as precursors in the synthesis of nucleic acids and so on. 4. Bacteria use oxygen substitutes such as nitrates. 	c. Pentose phosphate pathwad. Substrate-level phosphorylation
5. Pyruvic acid accepts electrons and is turned into various end-products, such as lactic acid or ethanol.	e. Anaerobic respirationf. Aerobic respiration
6. Glucose to pyruvic acid.	
IV. Matching	. Oxidativa nh canh awdation
 1. Electrons are removed from an organic compound and are transferred by an electron transport chain to oxygen. 2. An electron is liberated from chlorophyll and passes down 	a. Oxidative phosphorylationb. Substrate-level phosphorylation
an electron transport chain.	c. Photophosphorylation
V. Matching	
1. A dehydrogenase coenzyme derived from nicotinic acid (niacin).	a. NAD ⁺ b. Decarboxylation
2. A dehydrogenase coenzyme derived from riboflavin.	c. Coenzyme A
 3. In chemiosmosis, protons can diffuse across a membrane only through special channels that contain this enzyme. 	d. FMN
4. Pyruvic acid loses carbon dioxide to form an acetyl group.	e. ATP synthase
•	f. Dehydrogenation
VI. Matching	
1. Glycolysis.	a. Embden–Meyerhof
2. A photosynthetic organism that does not produce oxygen.	b. Chemoautotrophic
3. Removal of electrons.	c. Oxidation

d. Reduction

e. Anoxygenic

4. Uses an inorganic source of energy such as ammonia or

elemental sulfur.

Fill in the Blanks

1.	A chemoheterotroph that lives on dead organic matter is called a	
	When an enzyme's active site is occupied at all times by substrate or product molecules, it is called	ed
3.	Cyanide is an example of a general type of inhibitor called	
4.	Sulfa drugs are an example of a type of inhibitor called	
5.	In phosphorylation, no oxygen or other inorganic final electron accepts required.	oto
6.	Cyanobacteria produce gas, just as do higher plants.	
7.	The amount of ATP yield from aerobic respiration by a prokaryote is	
	The amount of ATP yield from glycolysis is	
9.	The removal of NH ₂ from an amino acid is called	
10.	The removal of —COOH from an amino acid is called	
11.	The substance acted upon by an enzyme is called the	
l2.	Coenzyme A is a derivative of the B vitamin acid.	
l3.	A sequence of enzymatically catalyzed chemical reactions in a cell is called a pathway.	
4.	Glucose is usually broken down to pyruvic acid by	
5.	In aerobic respiration, pyruvic acid is converted to acetyl; this produc can then enter the Krebs cycle.	ţ.
.6.	DNA and RNA are made up of repeating units called	

Critical Thinking

1. Why are catabolic and anabolic reactions referred to as coupled reactions?

2. Explain how competitive and noncompetitive enzyme inhibitors work.

3. How does the ultimate fate of electrons liberated differ in cyclic and noncyclic photophosphorylation?

4. What are the key features of the pentose phosphate pathway?